

Composition of Concentrated Reagent Grade Acids, Ammonium Hydroxide,

and Sodium and Potassium Hydroxide Solutions

Concentrated	Chemical	Formula	Approximate	ACS Assay	Molarity	Quantity to	Normality	Quantity to
Reagent	Formula	Weight	Strength	Limits (%w/w)	(M)	Prepare 1L of 1M	(N)	Prepare 1L of 1N
			(%w/w)			Solution (mL)		Solution (mL)
Acetic Acid	CH₃COOH	60.05	99.8	≥99.7%	17.4	57.5	17.4	57.5
Formic Acid	НСООН	46.03	90.0	88.0-96.0%	23.6	42.5	23.6	42.5
Hydrochloric Acid	HCI	36.46	37.2	36.5-38.0%	12.1	82.5	12.1	82.5
Hydrofluoric Acid	HF	20.01	49.0	48.0-51.0%	28.9	34.5	28.9	34.5
Nitric Acid	HNO ₃	63.01	70.0	68.0-70.0%	15.9	63.0	15.9	63.0
Nitric Acid	HNO₃	63.01	90.0	≥90.0%	N/A	N/A	N/A	N/A
(Fuming)								
Perchloric Acid	HCIO ₄	100.46	61.3	60.0-62.0%	11.7	85.5	11.7	85.5
Perchloric Acid	HCIO ₄	100.46	70.5	69.0-72.0%	9.5	105.5	9.5	105.5
Phosphoric Acid	H_3PO_4	98.00	85.5	≥85.0%	14.8	67.5	44.4	22.5
Sulfuric Acid	H_2SO_4	98.07	96.0	95.0-98.0%	18.0	55.5	36.0	28.0
Ammonium	NH₄OH	35.05	29.0 as NH ₃	28.0-30.0% as	14.5	69.0	14.5	69.0
Hydroxide				NH₃				
Potassium	КОН	56.11	45.0 as a	≥85% as pellets	11.7	85.5	11.7	85.5
Hydroxide			solution	45.0-46.0% as				
				solution				
Sodium	NaOH	40.00	50.5 as a	≥97% as pellets	19.4	51.5	19.4	51.5
Hydroxide			solution	50.0-52.0% as				
				solution				

This table is based on values established by the ACS on reagent Chemicals.